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institutionalized once the seed money disappears [Bacchetti and Ehrlich, 2006]. We believe that changes prompted by the CID will stick because these were choices made by the department itself, rather than as a response to an external agent" (page 165). As a dean I view this as the "if they really want to do it, they can do it on their own and thus they don't need the college's scarce resources" approach.

On the other hand, there is, for example, the approach of the NSF Division of Mathematical Sciences through its workforce program. So where does the truth lie?

Of course the truth is somewhere in between. I agree that reform needs to be a grassroots movement from faculty, staff, and students as a fundamental cultural shift is necessary. But oftentimes these shifts require more than just will. So how do you convince your administration or the NSF? I think the answer is similar for both. Further, most deans think alike on this point and it is no secret—but it is amazing to me at least—how few department chairs provide us the information we want/need to invest scarce resources in a department. We are looking for evidence that you have done your homework, that you have reflected on the issues, that you have well-thought-out plans, and that you have ways to evaluate the success of these plans if you are able to implement them. We are also looking for evidence that you have done all that you can do on your own. Everyone could use more resources, but what have you done to conserve the resources that you have? How high a priority is this to your department, as evidenced by a serious investment of your own time and resources? These issues can require serious and often delicate discussions, but progress is made not just by adding on but sometimes also by cutting back.

I'll step off my soapbox now and return to the book at hand. I highly recommend it to all faculty members interested in improving their graduate programs. It is a relatively quick read. Indeed, portions most relevant to other disciplines can often be skipped, although the cross-cultural enlightenment these sections can provide can be both amusing and worthwhile. The authors and the Carnegie Foundation for the Advancement of Teaching are to be commended.

### References

- HYMAN BASS, "The Carnegie Initiative on the Doctorate: The case of mathematics", *Notices Amer. Math. Soc.* 50, no. 7 (2003), 767–776.
- [2] TONY F. CHAN, "The mathematics doctorate: A time for change?", *Notices Amer. Math. Soc.* 50, no. 8 (2003), 897–904.
- [3] ALLYN JACKSON, The Carnegie Initiative on the Doctorate, Notices Amer. Math Soc. 50, no. 5 (2003), 566–568.

### About the Cover

#### Real analysis in polar coordinates

The cover image shows Ken Golden (author of the article on the mathematics of sea ice in this issue) measuring the fluid permeability of first year Antarctic sea ice. He writes about it:

"The fluid permeability of porous sea ice controls brine drainage, melt pond evolution, surface flooding, and snow-ice formation. Melt pond evolution, for example, in turn controls sea ice reflectance, a key parameter in understanding the dramatic decline of the summer Arctic ice pack. Snow-ice formation, a dominant process in the Antarctic, may well become more prevalent in the Arctic as the ice pack thins. The permeability also controls microbial colonization and nutrient fluxes, which are important in gauging the response of polar ecosystems to climate change. The permeability measurements I made on this expedition were the first such measurements done in the Antarctic ice pack. The photograph was taken by Jan Lieser."

> -Bill Casselman, Graphics Editor (notices-covers@ams.org)



# Climate Change and the Mathematics of Transport in Sea Ice

Kenneth M. Golden

s the boundary layer between the ocean and atmosphere in the polar regions, sea ice is a critical component of the global climate system. As tempera--tures on Earth have warmed, the Arctic sea ice pack in particular has exhibited a dramatic decline in its summer extent. Indeed, the polar sea ice packs are harbingers of climate change. Predicting what may happen over the next ten, fifty, or one hundred years requires extensive modeling of critical sea ice processes and the role that sea ice plays in global climate. Currently, large-scale climate models in general do not realistically treat a number of sea ice processes that can significantly affect predictions. Moreover, monitoring the state of Earth's sea ice packs, in particular their thickness distributions, provides key data on the impact of global warming. Mathematics is currently playing an important role in addressing these fundamental issues and will likely play an even greater role in the future. Here we give a brief bullet-by-bullet summary of the contents of this article.

### Question 1. What is the role of polar sea ice in global climate?

- Sea ice forms the boundary between the polar oceans and the atmosphere, and mediates the exchange of heat, gases, and momentum between them. The polar sea ice packs help regulate Earth's climate and are acute indicators of climate change.
- While sea water absorbs most incident solar radiation, sea ice reflects most of

it. Earth's sea ice packs act as solar heat reflectors, as well as ocean insulators, keeping significant heat from escaping to the atmosphere.

- Sea ice is a porous composite of pure ice with liquid brine inclusions. The flow of fluids through sea ice mediates processes important to climate, from melt pond evolution, which controls the reflectance of the sea ice packs, to upward percolation of sea water, which floods the surface and then freezes, forming *snow-ice*, an increasingly important component in the sea ice system. Brine flow though sea ice can also enhance heat exchange.
- Sea ice hosts algal and bacterial communities which support life in the polar oceans. Nutrient replenishment processes are controlled by fluid transport through the microstructure.
- The state of the polar sea ice packs—their extent and thickness—provides important information in monitoring climate change.

# Question 2. What is the role of mathematics in modeling transport in sea ice?

- Various techniques within the field of **homogenization** are used to derive macroscopic information about transport in sea ice from partial information about its microstructure.
- Variational formulations for the trapping constant and mean survival time for **diffusion processes** which interact with the pore boundaries are used to obtain rigorous bounds on the **fluid permeabilitytensor** for sea ice, based on general microstructural information.

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- Fluid transport in sea ice exhibits a permeable/impermeable transition at a critical brine volume fraction of about 5%, which controls geophysical and biological processes. **X-ray computed tomography** and mapping of the pore microstructure onto **random graphs** are used to demonstrate that the brine phase of sea ice undergoes a transition in connectedness at this brine volume fraction. **Percolation theory** is used to theoretically predict the transition and to mathematically characterize the thermal evolution of the fluid pores and their connectedness.
- Lattice and continuum percolation models are used to predict critical behavior of the fluid permeability in sea ice, with a universal exponent describing the behavior above the percolation threshold, and critical path analysis yields the scaling factor.
- **Hierarchical models** developed for porous rocks are used to predict the fluid permeability of sea ice over the entire range of brine porosities.
- A random pipe network, which is equivalent to a random resistor network and is solved using fast multigrid methods, is used to numerically simulate fluid flow through sea ice.
- Methods of **complex analysis** and **functional analysis** are used to obtain rigorous bounds on the **effective complex permittivity** of sea ice, the key parameter characterizing its electromagnetic behavior and the response of sea ice in remote sensing applications, such as monitoring sea ice thickness.
- **Inverse methods** yield microstructural information from complex permittivity data, paving the way for electromagnetic monitoring of fluid and thermal transport in sea ice.
- We have made **measurements of fluid and electrical transport** in sea ice on a 2007 Antarctic expedition in order to validate our models and investigate new phenomena.
- Large-scale sea ice and climate models currently do not generally incorporate the key processes our work on transport describes. These global models have grid sizes on the order of many kilometers. Future mathematical challenges include quantification of how local transport properties on the scale of individual floes influence pack behavior on much larger scales relevant to global models.

### Change in Arctic sea ice extent

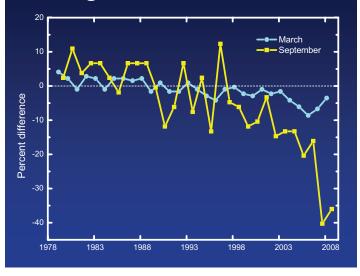


Figure 1. While winter Arctic sea ice extent (blue) has decreased modestly over the past thirty years, summer pack ice (yellow) has experienced a dramatic decline, particularly in the last couple of years. The dotted base line represents the average extent during the period 1979–2000. Some climate scientists wonder if this critical behavior of the system means we have passed through a so-called *tipping point* into a new regime.

This paper is divided into two main parts. First we discuss the role of sea ice in the climate system and how transport mediates key processes important to issues of climate change. Then we outline how mathematics is being used to model fluid and electromagnetic transport in sea ice. Only a small number of key references and starting points for further reading are given here. Many of the references for this work can be found in [8, 9, 10], and a Web link to an extensive bibliography is given at the end.

# Sea Ice and Its Role in the Global Climate System

### Decline of the summer Arctic sea ice pack

Up until about ten years ago, each summer most of the Arctic Ocean was covered with sea ice of areal extent roughly the same as the continental United States. By 2005, this thick, perennial ice pack had lost the areal equivalent of the U.S. east of the Mississippi River. By the end of the 2007 summer melt season, an area comparable to most of the Midwestern U.S. had been lost as well, and the Arctic ice pack had shrunk by nearly 40 percent from its 1979–2000 average extent. The fabled Northwest Passage was finally open to seafarers

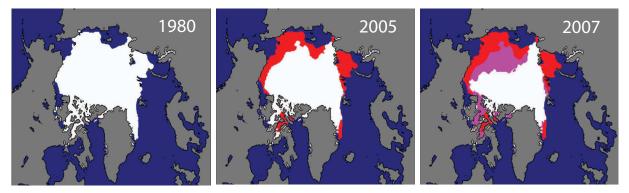


Figure 2. In 1980 the areal extent of the summer Arctic sea ice pack corresponded roughly to the continental U.S. By 2005 the area shown in red was gone, and by 2007 the additional area in purple was gone as well.

for the first time in human memory. These stunning losses in the extent of summer Arctic sea ice represent perhaps the most dramatic, large-scale indicator of global warming on the surface of the Earth so far.

The sea ice pack surrounding the continent of Antarctica is mostly seasonal, meaning it melts in the summer before returning in the cold, dark winter. Will the Arctic become mostly ice-free in summer like the Antarctic? If so, how long will it take? The answers to such questions are critical for global climate. The summer Arctic sea ice pack serves as a major part of Earth's polar refrigerator, cooling it and protecting the Arctic Ocean from absorbing too much solar heat. One of the central questions of climate science today is to understand why the Arctic ice pack has declined so precipitously and to predict how this complex system will evolve in the coming years. In order to see how mathematics can play a role in addressing such fundamental issues, let's first discuss how sea ice forms, explore its material structure, and then outline its role in global climate and polar ecosystems.

### Formation and structure of sea ice

Sea ice is frozen ocean water, which freezes at a temperature of about  $-1.8^{\circ}$ C, or  $28.8^{\circ}$ F. Initially, tiny platelike crystals called *frazil* form near the sea surface. Often they aggregate into slick patches called grease ice. When sea ice forms in calm conditions, thin sheets of translucent ice called nilas are seen first, which eventually thicken through congelation into gray ice and then white ice in large, undeformed floes. Currents or light winds often push the nilas around so that pieces slide over each other, a process known as rafting. Rafting nilas is shown in Figure 3 on top. In turbulent conditions typical of the open ocean, the "greasy" frazil suspension undergoes cyclic compression in the wave field, and during compression the crystals can freeze together

to form small cakes of slush. They grow larger by accretion and more solid through continued freezing between the crystals and form plates with raised rims known as *pancake ice* [22]. Patches of new grease ice are shown on the bottom in Figure 3. Pancake ice is shown in the bottom row of Figure 8. The pancakes in the lower right of Figure 8 are "glued" together in the process of forming larger floes. Sea ice that melts after one season is called *seasonal ice*, and sea ice that survives a summer season is called *multiyear ice*.

During the austral winter the continent of Antarctica, sitting atop the South Pole, is ringed mostly by seasonal ice on the Southern Ocean, with maximal ice extent in September. The Arctic is almost the geographical opposite, with an ocean at the North Pole, surrounded by continents, and covered with both seasonal and multiyear sea ice, of maximal extent in March. Sea ice should be contrasted with icebergs, glaciers, ice sheets, and ice shelves, which all originate on land. The great ice sheets covering Antarctica and Greenland are up to two miles thick and are composed of glacial ice formed from snow that has been compressed over thousands or millions of years into ice which flows under great pressure.

As a material, sea ice is quite different from glacial ice. When salt water freezes, the result is a polycrystalline composite of pure ice with inclusions of liquid brine, air pockets, and solid salts. As the temperature of sea ice increases, the porosity or relative volume fraction  $\phi$  of brine increases. Brine inclusions on the submillimeter scale are shown in the upper left panel of Figure 8, and its centimeter scale polycrystalline structure is shown in the upper right panel.

Sea ice exhibits a very interesting and important critical phenomenon [5]. For brine volume fractions  $\phi$  below about 5%, sea ice is effectively impermeable to fluid flow, while for  $\phi$  above 5%, it is increasingly permeable with  $\phi$ . This critical brine volume fraction  $\phi_c \approx 5\%$  corresponds to a critical temperature  $T_c \approx -5^{\circ}$ C for a typical bulk sea ice salinity of 5 parts per thousand, termed the *rule of fives*. When fluid flows through sea ice, transport is facilitated by *brine channels* connected brine structures ranging in scale from a few centimeters for horizontal slices to a meter or more in the vertical direction, as shown in the middle panel of Figure 8.

### Ice-ocean-atmosphere interactions and global climate

As the boundary layer between the ocean and atmosphere in the polar regions, sea ice is a key player in global climate and the world's ocean system and is a leading indicator of climate change [23, 12, 19, 17]. The sea ice pack mediates the exchange of heat, moisture, and momentum between the ocean and atmosphere, the two prinicpal geophysical fluids on Earth. For example, as winds push sea ice around on the ocean surface, momentum is transferred to the water below; the rougher the ice, the more efficient the transfer. This process can cause large-scale overturning in the upper ocean, bringing warmer water to the surface. As another example, when sea ice freezes, it rejects cold, dense brine, forming descending plumes in the polar oceans. In the Southern Ocean, this process leads to the formation of Antarctic bottom water, which then flows like a complex river through the world's oceans. Another important process in atmosphere-ocean interactions is the transfer of heat between them. In winter when the ocean is warmer, heat flows to the atmosphere through the sea ice itself, which forms an insulating blanket over the ocean, as well as through leads or openings in the pack. The thermal conductivity of sea ice is thus an important transport coefficient helping to quantify atmosphere-ocean interactions in coupled climate models.

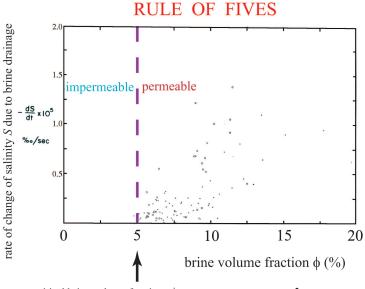
Roughly speaking, most of the solar radiation which is incident on snow-covered sea ice is reflected, while most of the solar radiation which is incident on darker sea water is absorbed. Sea ice is both ocean sunscreen and blanket, preventing solar rays from warming the waters beneath and thwarting ocean heat from escaping to warm the air above. The ratio of reflected sunlight to incident sunlight is called *albedo*. While the albedo of snow-covered ice is close to 1 (larger than 0.8), the albedo of sea water is close to zero (less than 0.1). If warming temperatures melt sea ice over time, fewer bright surfaces are available to reflect sunlight, more heat escapes from the ocean to warm the atmosphere, and the ice melts further. As more ice is melted, the albedo of the polar oceans decreases, leading to more solar absorption and warming, which in turn leads to more melting, in a positive feedback loop. It is believed that this so-called ice-albedo feedback has played



Figure 3. Sea ice forming in Antarctica, with nilas on top and grease ice on the bottom.

an important role in the marked decrease in Arctic sea ice extent in summer [16]. Thus, even a small increase in temperature can lead to greater warming over time, making the polar regions the most sensitive areas to climate change on Earth. Global warming is *amplified* in the polar regions. Indeed, global climate models consistently show amplified warming in the high latitude Arctic, although the magnitude varies considerably across different models. For example, the average surface air temperature at the North Pole by the end of the twenty-first century is predicted to rise by a factor of about 1.5 to 4 times the predicted increase in global average surface air temperature.

In the summer of 2007, the extent of the summer Arctic ice pack reached a record minimum. Ice mass balance observations show that there was an extraordinarily large amount of melting on the bottom of the ice in the Beaufort Sea during that summer [16]. Calculations indicate that solar heating of the upper ocean was the primary source for the observed melting. An increase in the open water fraction resulted in a 500% positive anomaly in solar heat input to the upper ocean, triggering an ice-albedo feedback, accelerating the pack ice retreat. Also of concern is that there has apparently been a significant reduction in the relative amount of thicker, multivear ice in the Arctic basin. Many authors have recently acknowledged that a relatively younger,



critical brine volume fraction  $\phi_c \approx 5\%$   $\longleftrightarrow$   $T_c \approx -5^{\circ}C, S \approx 5 \text{ ppt}$ 

Figure 4. Sea ice was observed in 1975 by Cox and Weeks (see [5]) to be effectively impermeable to brine drainage for  $\phi$  below about 5% and increasingly permeable with  $\phi$  above 5%.

thinner ice cover is more susceptible to the effects of atmospheric and oceanic forcing [17]. In the face of predictions for continued warming, the persistence of recent atmospheric and oceanic circulation patterns, and the amplification of these effects through the ice-albedo feedback mechanism, it is becoming increasingly likely that the Arctic Ocean will eventually change from perennially ice-covered to ice-free in the summer.

As evidenced from the previous discussion, a key determining factor in predicting the future trajectory of Arctic sea ice is understanding how ice pack albedo evolves. Melt ponds which form on the surface of the ice, as shown on the right in Figure 6, determine its albedo. Whether or not these ponds spread, deepen, or drain is controlled largely by the fluid permeability of the sea ice, which is one of the main motivations for our studies of fluid transport in sea ice discussed below.

Another motivation for studying fluid flow in sea ice is the importance of sea ice production from freezing of flooded ice surfaces, particularly in the Antarctic, where it is ubiquitous throughout the pack. As snow loading increases during a storm, a pressure head develops. Sea water is forced upward through the sea ice, as constrained by its permeability. Net flux to the surface can occur only if the temperature throughout the entire sea ice layer has exceeded the critical temperature  $T_c$  for fluid flow. Once the surface of the ice is flooded, the slushy mix of sea water and snow freezes into snow-ice, which is a dominant form of ice production in the Antarctic. In fact, witnessing this process during a powerful winter storm in the Eastern Weddell Sea while on the Antarctic Zone Flux Experiment (ANZFLUX) in 1994 and watching sea water flood the ice surface in a rather dramatic fashion were what motivated the author to begin studying fluid transport in sea ice. Recent models have even suggested that with increased polar precipitation accompanying global warming, the Antarctic winter sea ice pack could thicken and that snow-ice production could become more important in the Arctic. Fluid flow through sea ice also facilitates convection-enhanced thermal transport and the input of brine and fresh water into the upper ocean from freezing, melting, and drainage processes [23, 19].

### Sea ice ecology and fluid transport processes

The brine inclusions in sea ice host extensive algal and bacterial communities [14, 19] that are essential for supporting life in the polar oceans, such as krill, which themselves support fishes, penguins, seals, and Minke whales, and on up the food chain to top predators like killer whales, leopard seals, and polar bears. Nutrient replenishment processes for sea ice microbes are facilitated by fluid flow through the porous microstructure. For example, an algal bloom during the Antarctic fall in the Weddell Sea was fueled by nutrient-laden sea water replacing cold dense brine draining from freezing slush at the surface in a convective overturning process. The bloom effectively shut down, however, once the critical isotherm—above which the ice was impermeable and below which the ice was permeable—passed through the algae layer during the fall freeze-up [5].

### Multiscale structure of sea ice over ten orders of magnitude

One of the fascinating yet challenging aspects of modeling sea ice and its role in global climate is the sheer range of relevant length scales of structure, over ten orders of magnitude from the submillimeter scale to hundreds of kilometers. In Figures 8 and 9, we show principal examples of sea ice structure illustrating such a range of scales. Modeling sea ice on a large scale depends on understanding the physical properties of sea ice at the scale of individual floes and smaller. Today's climate models challenge even the most powerful supercomputers to their fullest capacity. However, even the largest computers still limit the horizontal resolution to tens of kilometers and require clever approximations to model the basic physics of sea ice. One of the central themes is how to use information at a smaller scale to predict behavior at a larger scale, such as in classical statistical mechanics. In the sections to follow we focus on the microphysics of sea ice. That is, we address the fundamental issues of characterizing the brine microstructure and its thermal evolution and how to use that knowledge to derive information about the bulk transport properties of sea ice relevant to climate models, as well as monitoring climate change.

### Bounds on the Fluid Permeability of Sea Ice

Despite the importance of fluid flow through sea ice in considerations of climate change and polar ecosystems, little attention had been paid to this problem, particularly from the theoretical standpoint, prior to our work [5, 9, 8, 26] (with [3] a notable exception). As a first step toward analyzing the vertical fluid permeability k and its microstructural dependence, we obtained rigorous upper bounds on  $k(\phi)$  [9, 8], which we briefly review.

Consider low Reynolds number flow of a fluid through a porous random medium, occupying a region  $\Omega \subset \mathbb{R}^3$ . Brine of viscosity  $\mu$  occupies the brine pore space  $\Omega_b \subset \Omega$ , having a relative volume fraction  $\phi$ , which is determined by the sea ice temperature *T* and bulk salinity *S* [23, 19]. The solid ice phase occupies the ice grain space  $\Omega_i \subset \Omega$ , having a relative volume fraction  $1 - \phi$ . Let  $\chi(x)$  be the indicator function of the brine pore space, with  $\chi(x) = 1$  if  $x \in \Omega_b$  and  $\chi(x) = 0$ if  $x \in \Omega_i$ . The porous microstructure is assumed to have translation invariant statistics for given *T* and *S* so that an *ergodic hypothesis* is satisfied, with equivalent ensemble and volume averaging (in the infinite volume limit), denoted by  $\langle \cdot \rangle$  [20].

As with many other porous media [20], there is a microscopic length scale  $\ell$  associated with the medium. For example, the scale over which the two-point correlation function obtained from  $\chi(x)$ varies is a good measure of this length, which is small compared to a typical macroscopic length scale L, such as a sample size or thickness of a statistically homogeneous layer. Then the parameter  $\epsilon = \ell/L$  is small, and one is interested in obtaining the effective fluid transport behavior in the limit as  $\epsilon \to 0$ . To obtain such information, the method of two-scale homogenization or twoscale convergence [20, 11] has been developed in various forms, based on the identification of two scales: a slow scale x and a fast scale  $y = x/\epsilon$ . The velocity and pressure fields in the brine,  $v^{\epsilon}(x)$ and  $p^{\epsilon}(x)$  with  $x \in \Omega_{h}$ , are assumed to depend on these two scales x and y. The idea is to average, or homogenize, over the fast scale using expansion in powers of  $\epsilon$ , which leads to a simpler equation describing the overall behavior of the flow.

The velocity and pressure in the brine phase satisfy the Stokes equations

(1) 
$$\nabla p^{\epsilon} = \mu \Delta v^{\epsilon}, \quad \nabla \cdot v^{\epsilon} = 0, \quad x \in \Omega_b,$$



Figure 5. The sea ice pack forms the boundary between the ocean and atmosphere in the polar regions. Here sea smoke forms in the track of the icebreaker *Aurora Australis* as heat flows from the warmer ocean to the colder atmosphere. Rougher ice on the right has a larger drag coefficient than smooth ice and transfers more momentum from the winds to the ocean below.

with the no-slip boundary condition on the pore surface  $v^{\epsilon}(x) = 0$ ,  $x \in \partial \Omega_b$ . The first equation in (1) is the steady state fluid momentum equation in the zero Reynolds number limit. The macroscopic equations can be derived through a two-scale expansion, and by averaging the leading order term  $v_0$  of the velocity over y, one obtains

(2) 
$$v(x) = -\frac{1}{\mu} \mathbf{k} \nabla p(x), \quad \nabla \cdot v(x) = 0,$$

where  $p(x) = p_0(x)$ , the leading order term in the expansion of  $p^{\epsilon}(x)$ , and **k** is the effective permeability tensor. The first equation in (2) is known as Darcy's law and the second is the macroscopic incompressibility condition [11]. We shall be interested in the permeability in the vertical direction  $k_{zz} = k$ , in units of m<sup>2</sup>.

To motivate our results, consider the effective conductivity tensor  $\boldsymbol{\sigma}^*$  of a two-phase composite with electrical conductivities  $\sigma_1$  and  $\sigma_2$  in the volume fractions  $\phi$  and  $1 - \phi$ . The defining equations here are  $J = \sigma E$ , where J is the current density, E is the electric field, and  $\sigma(x)$  is the local conductivity taking the values  $\sigma_1$  and  $\sigma_2$ . The local fields satisfy

(3) 
$$\nabla \times E = 0, \quad \nabla \cdot J = 0,$$

and the averaged equation  $\langle J \rangle = \boldsymbol{\sigma}^* \langle E \rangle$  defines  $\boldsymbol{\sigma}^*$ , analogous to Darcy's law. Optimal upper and lower bounds on a diagonal component  $\boldsymbol{\sigma}^*$  of  $\boldsymbol{\sigma}^*$ , such as in the vertical direction, known as the classical arithmetic and harmonic mean bounds



Figure 6. White, ice-covered polar regions are important for reflecting much of the incoming solar radiation during summer, whereas sea water absorbs most of the incident radiation. The reflectance, or albedo, of the sea ice pack is largely determined by melt ponds on the surface, like those shown on the right in the Arctic in June 2007. Melt pond evolution itself is largely determined by the fluid permeability of the sea ice below, as well as by the Sun's rays.



Figure 7. The rich Antarctic marine ecosystem is supported by photosynthetic algae (d), which give sea ice a brownish or yellowish color (e). Nutrient replenishment for these microbes is controlled by fluid flow through the porous brine microstructure. Krill (c) feed on algae, Emperor penguins (a) feed on krill, and killer whales (b) spy-hop looking for penguins to eat.

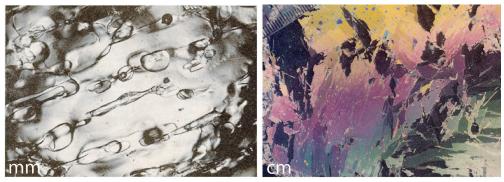
(or elementary bounds), were obtained almost one hundred years ago by Wiener:

(4) 
$$\frac{1}{\frac{\phi}{\sigma_1} + \frac{1-\phi}{\sigma_2}} \leq \sigma^* \leq \phi\sigma_1 + (1-\phi)\sigma_2.$$

The bounds are optimal, since laminates parallel to the field realize the upper bound, as shown in

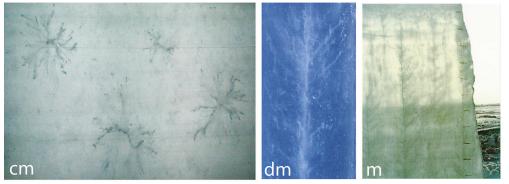
Figure 10(a), while laminates perpendicular to the field realize the lower bound.

For the fluid problem, we ask what configuration of the pores maximizes k. Intuitively, the best arrangement to maximize k is in vertical pipes of appropriate radii, as shown in Figure 10(b), where in the fluid problem a length scale not present for electrical conductivity arises. This



brine inclusions

polycrystals



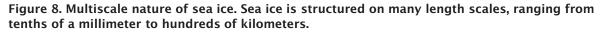
horizontal

vertical



brine channels

pancake ice



difference can be seen from the observation that the electrical conductance  $g_e$  of a cylinder of unit length and radius r is  $g_e = \pi r^2 \sigma$ , where  $\sigma$  is the conductivity of the material, while the fluid conductance  $g_f$  of a pipe of unit length and inner radius r is  $g_f = \pi r^4/8\mu$ . In [9] we obtained a *pipe bound*,  $k \le \phi [a(\phi)]^2/8$ , with mean brine inclusion horizontal cross-sectional radius  $a(\phi) = 7 \times 10^{-5} + (1.6 \times 10^{-4})\phi$  meters. The simple pipe bound, based on uniformly sized pipes whose radii increase with  $\phi$  or T, captures lab data [9]. However, when considering Arctic field data where a broader range of inclusion length scales is relevant, some of the data lie outside the bound. We must consider generalizations which allow for a distribution of radii. The derivation of such bounds [21, 9, 8] employs diffusion processes in the pore space.

Consider the problem of a tracer diffusing in the fluid phase  $\Omega_b \subset \Omega \subset \mathbb{R}^3$  of a porous medium. The tracer reacts with partially or completely absorbing *traps* on the boundary  $\partial \Omega_b$  of the pore space.

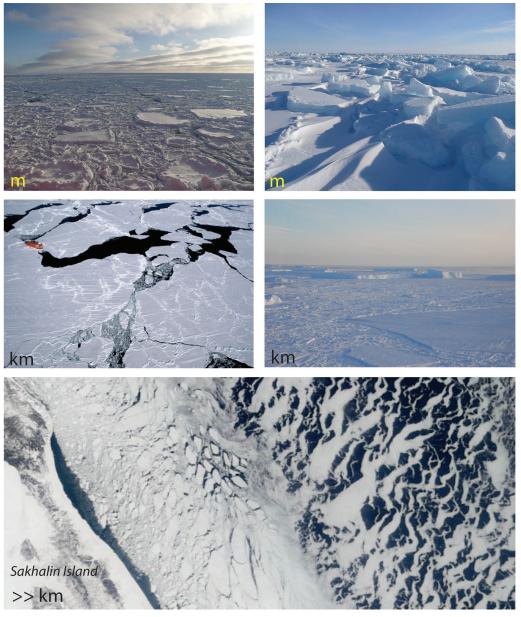


Figure 9. Multiscale nature of sea ice, continued. Pack ice viewed from ice level (top), from a helicopter (middle), and from a satellite (bottom).

(6)

Let c(x, t) with  $x \in \Omega_b$  be the time-dependent concentration of the reactant governed by the diffusion equation and boundary condition

(5) 
$$\frac{\partial c}{\partial t} = D\Delta c + G, x \in \Omega_b, D\frac{\partial c}{\partial n} + \eta c = 0, x \in \partial \Omega_b,$$

where *D* is the diffusion coefficient of the reactant in the fluid,  $\eta$  is a positive surface reaction rate constant, *G* is a generation rate of reactant per unit trap-free volume, and *n* is the unit outward normal from the pore space. In the diffusion-controlled limit [21], where a reactant will typically diffuse in the pore space for much longer than the characteristic time associated with the surface reaction, we consider the steady state problem where the rate of removal of the reactant by absorbing boundaries is exactly compensated by the production rate per unit volume G of the reactant.

The parameter of interest is the *trapping constant*,

$$\gamma^{-1} = \langle u\chi \rangle,$$

where *u* is a scaled concentration field satisfying  $\Delta u = -1$  in  $\Omega_b$  and u = 0 on the pore boundaries, obtained from a two-scale expansion of *c*. The trapping constant is inversely related to the *mean survival time*  $\tau$  for diffusion in the pore space,  $\tau = 1/\gamma \phi D$ . Using a variational definition

for the trapping constant with an appropriately constructed trial field yields a general two-point *void lower bound* [21]

(7) 
$$\gamma \ge \frac{(1-\phi)^2}{\ell_P^2},$$

where  $\ell_P$  is a length scale defined in terms of the two-point correlation function of the pores.

In the theory of composite materials, the Hashin-Shtrikman bounds [20, 9, 10] are important in providing optimal estimates on effective transport properties using the volume fraction information as well as isotropy of the material. These bounds are optimal in that they can be attained by actual geometries, namely, hierarchical arrays of coated spheres in d = 3 or coated cylinders in d = 2, filling all space. The two-point void lower bound in (7) can be evaluated [21] for Hashin-Shtrikman coated cylinder geometries, i.e., brine-filled vertical ice pipes for the sea ice problem, with appropriate distributions of inclusion radii  $R_I$ , as shown in Figure 11(a). Using an inequality relating the permeability tensor to the trapping constant [20] yields an optimal, rigorous upper bound,

(8) 
$$k \le \frac{\phi \langle R_I^4 \rangle}{8 \langle R_I^2 \rangle}$$

Measured brine inclusion cross-sectional areas *A* have been found to obey a lognormal distribution [8]. Then  $z = \ln A$  has a normal probability density with mean  $\alpha$  and variance  $\rho^2$ ,

(9) 
$$P(z) = \frac{1}{\sqrt{2\pi\rho^2}} e^{-(z-\alpha)^2/2\rho^2}.$$

We evaluated [8] the moments in (8) using the lognormal distribution, yielding

(10) 
$$k(\phi) \leq \frac{\phi}{8\pi} \langle A(\phi) \rangle e^{\rho^2}.$$

With variance  $\rho^2 \approx 1$  and  $\langle A(\phi) \rangle = \pi a^2(\phi)$  as above, the *lognormal pipe bound* in (10) captures our Arctic field data, as well as the lab data, as shown in Figure 11(b). The field data were taken near Barrow, Alaska, in a hydrological bail test, where a cylindrical hole is drilled, a tube is inserted to seal off horizontal flow, and time series measurements of the water level are used to estimate the permeability of the ice under the hole.

# Predictions for the Fluid Permeability of Sea Ice

### Percolation theory for sea ice

Percolation theory [18, 20, 11] has been used to model disordered materials where the connectedness of one phase dominates effective behavior. Examples range from porous rocks and bubbly glacial ice to semiconductors and carbon nanotube composites. Consider the square (d = 2) or cubic (d = 3) network of bonds joining nearest

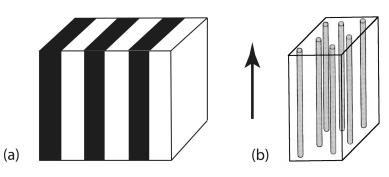


Figure 10. (a) The conductivity of a two-phase composite in a vertical electrical field is maximized by laminate structures parallel to the field. (b) The vertical fluid permeability of a porous medium is maximized by vertical pipes with appropriate radii.

neighbor sites on the integer lattice  $\mathbb{Z}^d$ . We will initially consider the problem of electrical transport through the network and then fluid transport. The bonds are assigned electrical conductivities  $\sigma_0 > 0$  (open) or 0 (closed) with probabilities p and 1 - p. Groups of connected open bonds are called open clusters, and the average cluster size grows as p increases. The striking feature of this model is that there is a critical probability  $p_c$ ,  $0 < p_c < 1$ , called the *percolation threshold*, where an infinite cluster of open bonds first appears. In d = 2,  $p_c = \frac{1}{2}$ , and in d = 3,  $p_c \approx 0.25$ .

The infinite cluster density  $P_{\infty}(p)$  is defined as the probability that the origin (or any point, by translation invariance) is contained in the infinite cluster, whose graph is shown in Figure 13(b). The two-point function  $\tau(x, y)$  measures the probability that two points x and y belong to the same open cluster. As  $|x - y| \rightarrow \infty$ ,  $\tau(x, y) \sim exp(-|x - y|/\xi)$ , where  $\xi$  is called the correlation length, which is a measure of the scale of connectedness. The correlation length diverges as p increases toward  $p_c$  and the infinite cluster forms, as shown in Figure 13(a). In the neighborhood of  $p_c$ ,  $P_{\infty}(p)$  and  $\xi(p)$  exhibit power laws, with

(11) 
$$\begin{aligned} P_{\infty}(p) \sim (p-p_c)^{\beta}, \ p \to p_c^+, \\ \xi(p) \sim (p_c-p)^{-\nu}, \ p \to p_c^-, \end{aligned}$$

where  $\beta$  and  $\nu$  are *universal* critical exponents that are believed to depend only on dimension and not on the details of the lattice. In d = 3,  $\beta \approx 0.41$  and  $\nu \approx 0.88$  [18].

Let  $\sigma^*(p)$  be the effective conductivity of this random resistor network in the vertical direction, as defined by Kirchoff's laws in the infinite volume limit [4]. For  $p < p_c$ ,  $\sigma^*(p) = 0$ , as shown in Figure 13 (c). For  $p > p_c$  near the threshold,  $\sigma^*(p)$ exhibits power law behavior,

(12) 
$$\sigma^*(p) \sim \sigma_0 (p - p_c)^t, \quad p \to p_c^+,$$

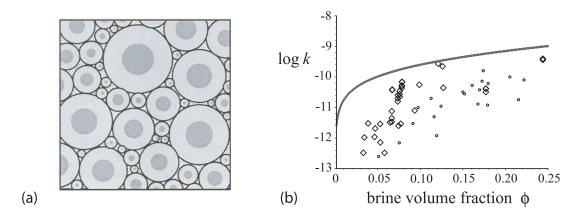


Figure 11. (a) Hashin-Shtrikman coated cylinders provide an optimal geometry which attains the two-point void upper bound for the vertical fluid permeability. (b) The lognormal pipe bounds capture all our lab and field data on the vertical permeability of sea ice.

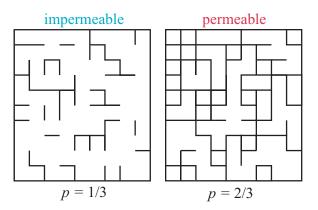


Figure 12. The two-dimensional square bond lattice below its percolation threshold  $p_c = 1/2$ in (a), and above its threshold in (b).

where *t* is the conductivity critical exponent. For lattices, *t* is believed to be universal, depending only on *d*. The corresponding fluid permeability  $\kappa^*(p)$  of the lattice, where the bonds are open or closed pipes of fluid conductivity  $\kappa_0/\mu = r_0^2/8\mu$ , behaves like

(13) 
$$\kappa^*(p) \sim \kappa_0 (p - p_c)^e$$
,  $p \to p_c^+$ ,

where *e* is the fluid permeability exponent. For lattices, it is believed that e = t [20]. In d = 2,  $t \approx 1.3$ , and in d = 3 [18, 20],  $t \approx 2.0$ . There is also a rigorous bound [4] that  $t \leq 2$ .

In view of the apparent critical behavior of brine drainage in columnar sea ice depending on whether  $\phi$  is above or below about 0.05 [23], it was reasonable to try and find a percolation theoretic explanation. However, with  $p_c \approx 0.25$  for the d = 3 cubic bond lattice, it was apparent that key features of the geometry of the brine microstructure in sea ice were being missed by lattices. In [5]  $\phi_c \approx 0.05$  was identified with the critical probability in a continuum percolation

model for compressed powders [13] that exhibit microstructural characteristics similar to sea ice. The identification explained the rule of fives, as well as data on algal growth and snow-ice production. The compressed powders, shown in Figure 14, were used in the development of so-called *stealthy* or radar-absorbing composites.

When we applied the compressed powder model to sea ice about a decade ago, we had no direct evidence that the brine microstructure undergoes a transition in connectedness at a critical brine volume fraction. This lack of evidence was due partly to the difficulty of imaging and quantitatively characterizing the brine inclusions in three dimensions, particularly the thermal evolution of their connectivity. Through X-ray computed tomography and pore structure analysis [8] we have now analyzed critical behavior of the thermal evolution of brine connectedness in sea ice single crystals, over a temperature range from  $-18^{\circ}$ C to  $-3^{\circ}$ C. We have mapped three-dimensional images of the brine phase onto graphs and analyzed their connectivity as functions of temperature and sample size. Realistic network models of brine inclusions can be derived from porous media analysis of 3-D microtomography images. Approaches include the well-established 3DMA package (3-Dimensional Medial Axis; see [8] for example), and the developmental tight dual model of Glantz and Hilpert. In both cases, a 3-D graph of nodes and edges is derived from the pores and throats. In general terms, the steps are: (1) represent the pore space by a polyhedral complex, (2) decompose this digital pore space into pores and throats and compute individual characteristic radii, (3) map these to nodes and edges tagged with their respective radii. The resulting graphs enable imbibition and drainage characterization, connectivity, and critical path analyses (see below).

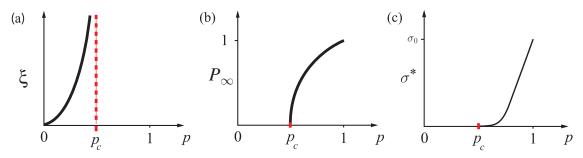


Figure 13. (a) The correlation length diverges as  $p \to p_c^-$ . (b) The infinite cluster density  $P_{\infty}(p)$  vanishes below  $p_c$  and takes off with power law behavior above  $p_c$ . (c) Below  $p_c$  there is no bulk transport, and above  $p_c$  the transport coefficient takes off with a different power law.

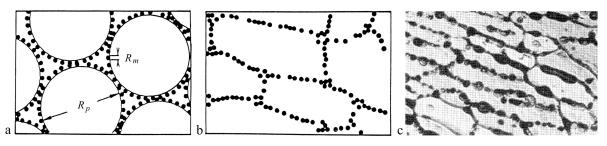


Figure 14. (a) A powder of large polymer spheres mixed with smaller metal spheres [13]. (b) When the powder is compressed, its microstructure is similar to that of sea ice in (c).

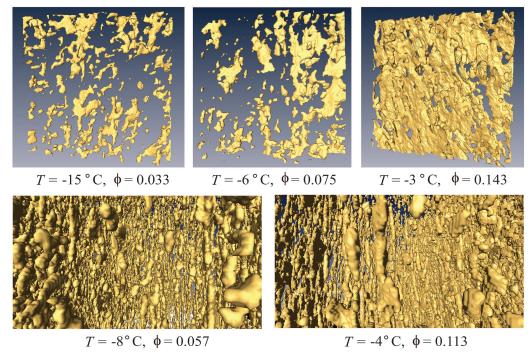


Figure 15. X-ray CT volume rendering of brine layers within a lab-grown sea-ice single crystal with S = 9.3 ppt. The (non-collocated)  $8 \times 8 \times 2$  mm sub-volumes (top row) illustrate a pronounced change in the micro-scale morphology and connectivity of the brine inclusions during warming. Images from a cylinder of height 8 mm and diameter 21 mm are shown in the bottom row.

An example of how brine morphology and connectivity evolve with temperature is shown in Figure 15. X-ray CT has provided new insight into small-scale, intracrystalline brine layer connectivity. Our work provides the first direct observation of an onset in connectivity over this scale ( $\approx$ 1 cm). Notably, the thermal evolution of pore morphology that we observe is more complex than the classic model [19, 23], a transition from brine sheets to cylindrical to spheroidal inclusions, which underlies a range of porosity models [19, 23].

While percolation theory has been applied to disordered composites for fifty years, key functions such as the percolation probability (which equals 0 for  $p < p_c$  and equals 1 for  $p > p_c$ ), the infinite cluster density, the correlation length, and their critical exponents have been computed only for idealized models, like the square or cubic lattice, and *not* for real materials. In general, it is difficult not only to image the microstructure of complex disordered media but to conveniently vary the volume fraction of the connected phase in which one is interested, such as the pore space in rocks or the doping in semiconductors. However, for sea ice, by changing the temperature of a single sample we can significantly vary the brine volume fraction. We have computed these functions from our three-dimensional imaging of the pore-space evolution (in a submitted paper by Pringle, Miner, Eicken, and Golden). For example, from analysis of the two-point connectivity function, we have generated a graph illustrating the divergence of the correlation length quite similar to Figure 13(a).

We have also used our data to find anisotropic percolation thresholds using *finite-size scaling*. Rather than attempting to compute very sensitive parameters such as the percolation threshold or critical exponents from real data and its variation in volume fraction, finite-size scaling is a technique from physics where one exploits the dependence of functions of interest on sample size to obtain such results. In our case, a key relation is

(14) 
$$P_{\infty}(p_c,L) \sim L^{-\beta/\nu}, \quad L \to \infty.$$

In single crystals of sea ice the brine microstructure is highly anisotropic, with inclusions usually arranged in layers. Using three different approaches, we have employed finite-size scaling to obtain anisotropic critical porosities of  $4.6 \pm 0.7\%$  in the vertical direction and, laterally,  $9 \pm 2\%$  parallel to the layers and  $14 \pm 4\%$  perpendicular to them. The value of about 4.6% in the vertical direction for single crystals should be smaller than the threshold of about 5% for bulk connectivity, which it is.

Now we consider the application of percolation theory to understanding the fluid permeability of sea ice. In the continuum [18, 20], the permeability and conductivity exponents e and t can take nonuniversal values and need not be equal, such

as for the three-dimensional Swiss cheese model. Continuum models have been studied by mapping to a lattice with a probability density  $\psi(g)$  of bond conductances g. Nonuniversal behavior can be obtained when  $\psi(g)$  is singular as  $g \to 0^+$ . However, for a lognormal conductance distribution arising from intersections of lognormally distributed inclusions, as in sea ice, the behavior is *universal* [8]. Thus  $e \approx 2$  for sea ice.

The permeability scaling factor  $k_0$  for sea ice, analogous to  $\kappa_0$  in (13), is estimated using critical path analysis (see the article by the author in [11]). For media with g in a wide range, the overall behavior is dominated by a critical, *bottleneck* conductance  $g_c$ , the smallest conductance such that the critical path  $\{g : g \ge g_c\}$  spans the sample. With most brine channel diameters [23] between 1.0 mm and 1.0 cm, spanning fluid paths have a smallest characteristic radius  $r_c \approx 0.5$  mm, and we estimate  $k_0$  by the pipe-flow result  $r_c^2/8$ . Thus

(15)  $k(\phi) \sim 3 (\phi - \phi_c)^2 \times 10^{-8} \text{ m}^2$ ,  $\phi \rightarrow \phi_c^+$ . In Figure 16(c), field data with  $\phi$  in [0.055, 0.15], just above  $\phi_c \approx 0.05$  [5], are compared with (15), showing close agreement. A best fit for *e* of 1.94 was obtained from other data. The striking result that for sea ice  $e \approx 2$ , the universal *lattice* value in three dimensions [20, 4], is due to the general lognormal structure of the brine inclusion distribution function. The general nature of our results suggests that similar types of porous media, such

as saline ice on extraterrestrial bodies, may also

#### Hierarchical models

exhibit universal critical behavior.

To model  $k(\phi)$  over all porosities, not just near the percolation threshold, we consider features of the brine phase present over the full rangesome degree of connectivity, particularly on small scales, and self-similarity. The basic unit of sea ice microstructure is an ice grain surrounded by brine and ice, having substructure itself on smaller scales. Hierarchical, self-similar models of spheres or other grains surrounded by smaller spheres, and so on, with brine of conductivity  $\sigma_w$  in the pore spaces, have been used to describe the transport properties of sedimentary rocks [24]. Such microstructures are similar to those shown in Figure 11(a) except in this case the basic elements are brine-coated spheres of ice. Effective medium theory gives Archie's law  $\sigma = \sigma_w \phi^m$  as the conductivity of the model with exact results for the exponent *m*, such as m = 3/2 for spherical grains. The corresponding permeability exponent [24] is 2m, so that with  $k_0$  as above and m = 3/2, the simplest hierarchical model yields

(16) 
$$k(\phi) = 3 \phi^3 \times 10^{-8} \text{ m}^2.$$

In Figure 17(a) the full *in situ* data set is compared to this theoretical prediction, with close

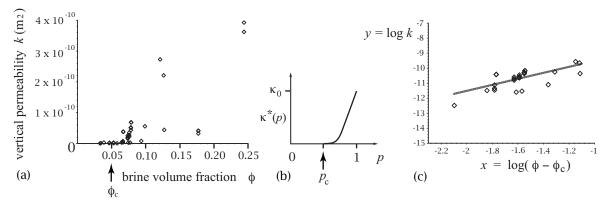


Figure 16. (a) Data for k taken *in situ* on Arctic sea ice, displayed on a linear scale. (b) The effective permeability  $\kappa^*(p)$  in a lattice percolation model, displayed on a linear scale. (c) Comparison of Arctic permeability data in the critical regime (25 data points) with percolation theory in (15). In logarithmic variables the predicted line has the equation y = 2x - 7.5, while a best fit of the data yields y = 2.07x - 7.45, assuming [5]  $\phi_c = 0.05$ .

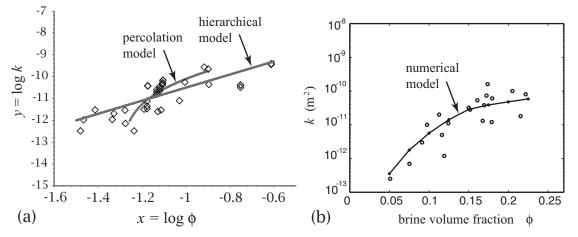


Figure 17. (a) Comparison of Arctic permeability data (37 data points) with our theoretical prediction in (16). In logarithmic variables the predicted line has the equation y = 3x - 7.5, while a best fit of the data yields y = 3.05x - 7.50. Lab data on artificially grown sea ice [3] give a best fit of y = 3.1x - 7.7 [8]. (b) Comparison of lab data with numerical simulation.

agreement. From photomicrographs of marine sand microstructures, compacted platey sand appears the most similar to the cellular substructure of sea ice. In this case,  $2m \approx 3.04$ , which is closer to the best fit of 3.05 from our field data. For calcite aggregates and Fontainebleau sandstone, a cubic power law like (16) agrees closely with permeability data for porosities above about 10%, with an exponent  $2m \approx 3.05$  for Fontainebleau sandstone.

Our results show that the parameters characterizing fluid transport in sea ice are almost the same as in some crustal, sedimentary rocks. Both types of media exhibit critical behavior near a percolation threshold, with  $\phi_c \approx 5\%$  for sea ice, and  $\phi_c \approx 4\%$  for calcite aggregates and Fontainebleau sandstone. Moreover, even though the porous microstructures in these rocks are quite different from sea ice, the permeability critical exponent in both cases falls into the lattice universality class. The permeability exponent of 3 for the hierarchical model is also about the same in both cases. It is interesting to note that a variation of four orders of magnitude in the permeability of calcite aggregates is achieved with temperatures ranging from 633 K to 833 K, and with confining pressures from 200 MPa to 300 MPa. For sea ice, a similar variation in permeability occurs over a temperature range of about 15 K.

### Numerical model for fluid flow through sea ice

In [26] we developed a two-dimensional random pipe network to simulate fluid flow through a porous medium, as illustrated in Figure 18. For

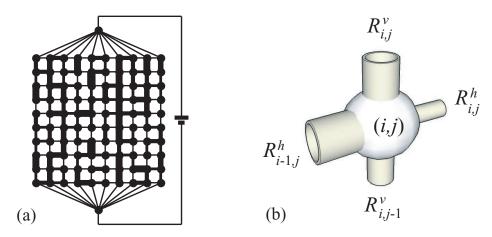


Figure 18. (a) Random pipe or resistor network. (b) Close-up of a node and adjoining ducts.

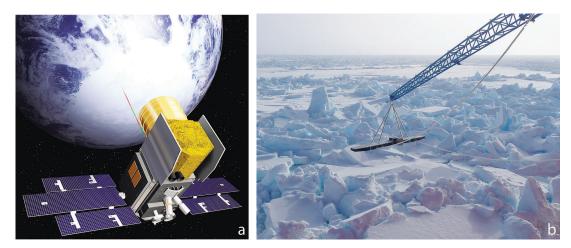


Figure 19. (a) NASA's Ice, Cloud and Land Elevation Satellite (ICESat) over Antarctica. It bounces laser pulses off the surfaces of ice sheets and sea ice to measure changes in thickness and monitor climate change. (b) An electromagnetic induction device for measuring sea ice thickness, called the "Worbot" after Australian sea ice scientist Tony Worby. It is mounted here on the icebreaker *Aurora Australis* in the Antarctic sea ice pack, September 2007.

sea ice the cross-sectional areas *A* of the pipes are chosen from the lognormal distribution in (9), with  $\langle A(\phi) \rangle$  as before and  $\rho^2 = 1$ . The system is equivalent to a resistor network, which was solved using a fast multigrid method. While the model compared well with lab data for  $\phi > 0.15$ , it overestimated  $k(\phi)$  for  $\phi < 0.15$ , where percolation effects, which were ignored, become significant. In accordance with our X-ray CT data, we introduced random *disconnections* so that the network falls apart as  $\phi$  decreases. The results then agree well with the lab data, as shown in Figure 17(b).

As discussed above, heat is exchanged between the ocean and atmosphere in the polar regions through sea ice. The thermal conductivity of sea ice is thus an important transport coefficient entering into climate models. We will not discuss our work in this area, but simply mention that the methods for studying transport we have outlined here can be applied to bounding and numerically calculating this important sea ice parameter. One particularly interesting aspect of the thermal transport problem is that when brine flows through sea ice, such as for temperatures above the percolation threshold, there is an advective term in the heat equation. A velocity field helping to move heat around can enhance measured thermal conductivity, which has been observed in Antarctic sea ice. In conjunction with our fluid permeability work, we have been theoretically and numerically modeling such processes.

### Sea Ice Remote Sensing: Monitoring the Impact of Climate Change

Because of the global nature of monitoring Earth's sea ice packs, information is usually obtained

via remote sensing from platforms on satellites, aircraft, and ships [15, 2, 7, 6]. The physics underlying the problem of remotely sensing sea ice concerns the interaction of electromagnetic waves with a polycrystalline composite of four components: pure ice, brine inclusions, air pockets, and solid salt deposits [23, 19]. Here we will focus on the two main constituents in sea ice, namely, pure ice with brine inclusions. One of the grand challenges of sea ice remote sensing is to accurately recover the thickness distribution of the pack. Assessing the impact of global warming on the polar regions involves monitoring not only the ice extent but the ice volume, which requires knowledge of ice thickness. While microwaves in the GHz frequency range have been commonly used in many applications, due to the absorbing nature of the brine phase and the short wavelengths involved, they typically can penetrate only into the upper portions of a sea ice floe. Lasers, such as on ICESat, illustrated in Figure 19(a), have been used to bounce off the ice or snow surface, and when compared with returns from the sea surface in open leads, they can provide thickness estimates. Changes in ice sheet elevation in Antarctica and Greenland can also be similarly obtained.

Recently there has been increasing interest in using low-frequency electromagnetic induction devices to estimate sea ice thickness [25], as shown in Figure 19(b). Electromagnetic fields in the MHz range, with wavelengths on the scale of meters, are used to probe the air-ice-ocean interface and estimate the thickness of the ice layer. The technique relies on a time-varying primary magnetic field (generated by a transmitter coil) inducing eddy currents in the seawater beneath the comparatively resistive ice. The secondary magnetic field produced is sensed by a receiver coil, determining an apparent conductivity that results essentially from an integration over the vertical distance between the instrument and induced currents. Accurate interpretation of the data relies on understanding how the conductivity of the sea ice, or more generally its complex permittivity, depends on the microstructural characteristics. For example, current methods of obtaining thickness estimates assume a single value for the conductivity. However, this key parameter can vary by over two orders of magnitude with depth throughout a given floe. Since typical sizes of the brine inclusions are on the submillimeter scale, the electromagnetic behavior of the sea ice can be treated using the quasi-static approximation, where the wavelength is much longer than the microstructural scale. Then the electromagnetic response of sea ice can be characterized by its effective complex permittivity tensor  $\epsilon^*$ . For a given diagonal component in a particular direction, the real part corresponds to the polarizability, or dielectric constant, in that direction, and the imaginary part corresponds to the conductivity in that direction. For example, the reflection coefficient of a wave incident on a flat sea ice surface is determined by the mismatch in complex permittivity between the sea ice and the air above.

There has been considerable work in the past on estimating and bounding  $\boldsymbol{\epsilon}^*$  for sea ice, particularly in the microwave region [2, 7]. We will discuss here some of our work on bounding  $\boldsymbol{\epsilon}^*$ , particularly in this low-frequency range, although the mathematics is quite general. We will also discuss how we've been inverting these bounds and other information to recover microstructural characteristics, which is important in monitoring fluid and thermal transport in sea ice. In this connection, there are new methods, such as crossborehole tomography developed by M. Ingham and colleagues, which show promise in reconstructing the permittivity profile of sea ice and, in turn, the state of the ice itself.

Here we review our work on theoretical bounds on the principal components of  $\epsilon^*$  with extensive outdoor tank and *in situ* data on  $\epsilon^*$  at 50 MHz, consisting of thousands of measurements taken with a capacitance probe [10]. In particular, we compare the low-frequency measurements with the Bergman-Milton bounds [7, 10] for the complex permittivity of a two-phase composite with known constituent permittivities  $\epsilon_1$  and  $\epsilon_2$  and volume fractions  $p_1$  and  $p_2$ . There are two types of these bounds which we employ. The first type assumes only the permittivity and volume fraction information and is known as the complex elementary or arithmetic and harmonic mean bounds. The second type further assumes statistical isotropy in the composite microstructure and is known as the complex Hashin-Shtrikman bounds.

All the data considered are situated well inside the complex elementary bounds. However, there are naturally occurring anisotropies in the brine microstructure, such as its preferred vertical orientation [23, 19], which become more pronounced above the critical brine volume fraction of about 5%, where the brine phase percolates [5]. Also, preferred azimuthal alignment of the c-axes of individual crystals within the horizontal plane has long been observed in the Arctic and was present in the Arctic sea ice whose permittivity was measured in one of the data sets considered here. Preferred c-axis alignment is attended by anistropy in the brine microstructure and anisotropic behavior of the complex permittivity within the horizontal plane. Comparing the different data sets with the isotropic bounds yields interesting and useful insights about sea ice microstructure, its evolution with temperature, and its electromagnetic properties. We also invert the complex bounds to obtain rigorous estimates of the brine volume fraction from the permittivity data [6].

#### Bounds on the complex permittivity of sea ice

Let us briefly describe the analytic continuation method for studying the effective properties of composite materials [7, 10]. Let the local complex permittivity  $\epsilon(x)$  be a spatially stationary random field in  $x \in \mathbb{R}^d$  for an appropriate probability space representing the set of realizations of the random medium. We assume  $\epsilon(x)$  for 50 MHz takes the values  $\epsilon_1 = 63.3 + i1930$  in brine and  $\epsilon_2 = 3.06$  in ice, and write  $\epsilon = \epsilon_1 \chi + \epsilon_2(1 - \chi)$ , where  $\chi(x)$  is again the characteristic function of the brine phase. Let E(x) and D(x) be the stationary random electric and displacement fields satisfying the constitutive law  $D(x) = \epsilon(x)E(x)$  and the equations

(17) 
$$\nabla \times E(x) = 0, \quad \nabla \cdot D(x) = 0,$$

with  $\langle E(x) \rangle = e_k$ , where  $e_k$  is a unit vector in the  $k^{th}$  direction for some k = 1, ..., d, and  $\langle \cdot \rangle$  means an ensemble average or spatial average over all of  $\mathbb{R}^d$ .

The effective complex permittivity tensor  $\boldsymbol{\epsilon}^*$  is defined by

(18) 
$$\langle D \rangle = \boldsymbol{\epsilon}^* \langle E \rangle.$$

For simplicity, we focus on one diagonal coefficient,  $\epsilon^* = \epsilon_{kk}^*$ . Due to the homogeneity of effective parameters,  $\epsilon^*(\lambda\epsilon_1,\lambda\epsilon_2) = \lambda\epsilon^*(\epsilon_1,\epsilon_2)$ ,  $\epsilon^*$  depends only on the ratio  $h = \epsilon_1/\epsilon_2$ , and we define  $m(h) = \epsilon^*/\epsilon_2$ . The two main properties of m(h) are that it is analytic off  $(-\infty, 0]$  in the *h*-plane and that it maps the upper half-plane to the upper half-plane, so that it is an example of a Herglotz, or Stieltjes, function. The key step in the analytic continuation method is obtaining an integral representation for  $\epsilon^*$ .

It is more convenient to work with the function F(s) = 1 - m(h), where s = 1/(1 - h), which is analytic off [0, 1] in the *s*-plane. It was proven that F(s) has the powerful representation

(19) 
$$F(s) = \int_0^1 \frac{d\mu(z)}{s-z}, \quad s \notin [0,1],$$

where  $\mu$  is a positive measure on [0,1] (not to be confused with the viscosity from before). Formula (19) separates the parameter information in *s* from information about the mixture geometry contained in  $\mu$ , a spectral measure of the operator  $\Gamma \chi$ , where  $\Gamma = \nabla (-\Delta)^{-1} \nabla \cdot .$  Statistical assumptions about the geometry are incorporated into  $\mu$  via its moments  $\mu_n = \int_0^1 z^n d\mu(z)$ , which can be calculated from the correlation functions of the random medium, with  $\mu_n = (-1)^n \langle \chi[(\Gamma \chi)^n e_k] \cdot e_k \rangle$ .

Bounds on  $\epsilon^*$ , or F(s), are obtained by fixing s in (19), varying over admissible measures  $\mu$  (or admissible geometries), such as those that satisfy only  $\mu_0 = p_1$ , and finding the corresponding range of values of F(s) in the complex plane. Two types of bounds on  $\epsilon^*$  are obtained. The first bound  $R_1$  assumes only that the relative volume fractions  $p_1$  and  $p_2 = 1 - p_1$  of the brine and ice are known,

with  $\mu_0 = p_1$  satisfied. In this case, the admissible set of measures forms a compact, convex set  $\mathcal{M}_0$ . Since (19) is a linear functional of  $\mu$ , the extreme values of F are attained by extreme points of  $\mathcal{M}_0$ , which are the Dirac point measures  $p_1 \delta_z$ . The values of *F* lie inside the region  $R_1$  bounded by circular arcs, called the complex elementary bounds. The formulas can be found in [7, 10]. This region collapses to the interval  $(p_1/\epsilon_1 + p_2/\epsilon_2)^{-1} \leq$  $\epsilon^* \leq p_1 \epsilon_1 + p_2 \epsilon_2$  when  $\epsilon_1$  and  $\epsilon_2$  are real, which are the arithmetic (upper) and harmonic (lower) mean bounds. The complex elementary bounds are optimal and can be attained by coated ellipsoidal geometries as the aspect ratio varies. If the material is further assumed to be statistically isotropic, i.e.,  $\epsilon_{ik}^* = \epsilon^* \delta_{ik}$ , then  $\mu_1 = p_1 p_2/d$  must be satisfied as well. A smaller bounding region  $R_2$  is then obtained, which forms the complex Hashin-Shtrikman bounds. Extensions of the analytic continuation method and bounds to multicomponent media involve techniques of several complex variables.

Now we compare three extensive sets of data on the complex permittivity of sea ice with the bounds. Measurements were taken in land-fast ice in the Chukchi Sea near Barrow, Alaska (14,004 data points), and in McMurdo Sound, Antarctica (2,382 data points), and in artificial, young sea ice in an outdoor tank in Fairbanks, Alaska (6,403 data points). The permittivity was measured by deploying an array of Stevens Water Monitoring Systems Hydraprobes. Each Hydraprobe is a coaxial probe with a central tine surrounded by three equally spaced outer tines, aligned horizontally, as shown in the photo inset in Figure 20(a). The three outer tines are held at ground potential, and a voltage is applied to the central tine at 50 MHz frequency, resulting in a circularly polarized wave with electric field predominantly in the plane perpendicular to the tines. The complex permittivity measured by the probe represents an average of the components of  $\boldsymbol{\epsilon}^*$  in this plane.

In Figure 20(a) the full data set for the Fairbanks ice tank experiment is displayed along with a series of regions  $R_1(\phi)$  for the indicated values of the brine porosity  $\phi = p_1$ . The data have been grouped into intervals of porosity—[0.02, 0.03], [0.03, 0.04], ..., [0.08, 0.09]—with a different color for each interval. The corresponding elementary bound has the same color. All the data in each interval lie inside the corresponding bound. The data marked with the gray color lie in the porosity interval [0.049, 0.051], which corresponds to the percolation threshold of about 5%, conjectured in [5].

It is useful to note that tighter bounds [7] on  $\epsilon^*$  can be obtained if the material is known to have a matrix-particle structure with separated inclusions, like sea ice at temperatures colder than its critical temperature. In this case, the support

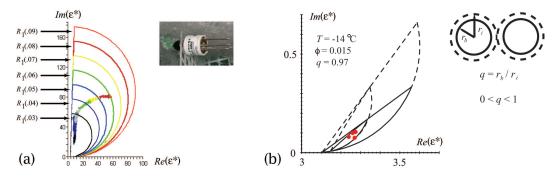


Figure 20. (a) Comparison of data on  $\epsilon^*$  at 50 MHz for sea ice grown in an ice tank with the elementary bounds, showing excellent correspondence. Data in a brine porosity interval signified by a given color lie inside the region of the same color. The data colored gray have porosities in the interval [0.049,0.051], close to the percolation threshold. The capacitance probe is shown on the right. (b) Matrix-particle bounds on  $\epsilon^*$  (solid curves), along with the complex elementary and Hashin-Shtrikman bounds (dotted curves), compared with 4.75 GHz data in red.

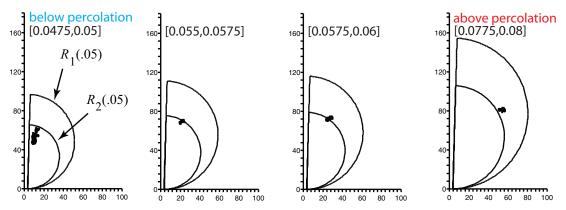


Figure 21. Comparison of ice tank data on the complex permittivity of sea ice at 50 MHz over ranges of  $\phi$  with the elementary (outer) and isotropic (inner) bounds. The data violate, or lie outside, the isotropic bounds as brine volume increases through the percolation threshold and an anisotropic, connected brine structure is formed.

of  $\mu$  lies in an interval  $[s_m, s_M]$ ,  $0 < s_m < s_M < 1$ . The more separated the inclusions, the smaller the support interval and the tighter the bounds. Comparison of higher-frequency 4.75 GHz data (red dots) on the complex permittivity  $\epsilon^*$  of sea ice for  $\epsilon_1 = 51 + i45$  and  $\epsilon_2 = 3.1$ , with a series of bounds [7], is shown in Figure 20(b). The outer dotted curves are complex arithmetic and harmonic mean bounds, and the inner dotted curves are Hashin-Shtrikman bounds. The solid bounds are matrix-particle versions. The parameter *q* measures inclusion separation and can be bounded using permittivity data.

In Figure 21 we display a series of comparisons of the 50 MHz data in the porosity intervals shown, with corresponding elementary bounds  $R_1(\phi_{max})$  and the isotropic bounds  $R_2(\phi_{max})$ , where  $\phi_{max}$  is the largest porosity in each interval. The development of anisotropy in the dielectric measurements, or violation of the isotropic bounds as brine volume increases through 5%, is consistent with an

increasingly connected and vertically elongated brine microstructure. This behavior in the data would likely be accentuated if the electric field were oriented predominantly in the vertical direction.

#### Inverse bounds for structural parameters

The objective of inverse bounds is to use data about the electromagnetic response of sea ice to recover information about its structural parameters, in particular its brine volume fraction  $\phi$ , or porosity. As mentioned above, the porosity  $\phi$  can be written as a function of the bulk ice salinity *S* and temperature *T*, as derived by semi-empirical approaches from the sea ice phase relations. Given salinity information, for example, inverting for porosity yields temperature information. The inverse method [6] we use here yields intervals of uncertainty for the brine volume fraction  $\phi$ . Given an observed value of the complex permittivity

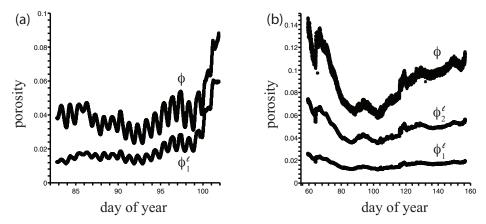


Figure 22. Comparison of the actual brine porosity  $\phi$  with time to the inverse lower bounds for the ice tank data in (a) and for the Barrow data in (b).

 $\epsilon^*$ ,  $\phi$  is increased until the value of  $\epsilon^*$  touches one boundary of the region  $R_1$  described in the previous section and is then decreased until the value touches the other boundary. This procedure gives a range of values  $\phi_1^{\ell} \le \phi \le \phi_1^{u}$ , with

(20) 
$$\phi_1^{\ell} = |f|^2 \frac{Im(\overline{s})}{Im(f)}, \qquad \phi_1^u = 1 - \frac{|g|^2 Im(\overline{t})}{Im(g)},$$

where *f* is the known value of *F*(*s*) and *g* is the known value of  $G(t) = 1 - \epsilon^*/\epsilon_1$  with t = 1 - s. If the material is further assumed to be statistically isotropic, second-order inverse bounds  $\phi_2^{\ell} \le \phi \le \phi_2^{u}$  can be obtained. We mention recent progress by E. Cherkaev and colleagues on developing methods for reconstructing  $\mu$  from bulk property measurements. Such methods will likely be useful in obtaining more detailed information about sea ice microstructure, as they have been for bone microstructure and in monitoring osteoporosis [1].

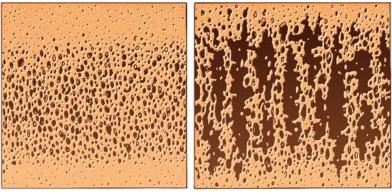
In Figure 22(a) the actual brine porosity  $\phi(t)$  as a function of time *t* for the ice tank experiment is compared with the lower inverse bound  $\phi_1^{\ell}$ . Due to the high contrast in the materials, the corresponding upper bound is quite large and currently provides little information. We see that the actual data lie above the lower bound and that the variations in the reconstructed lower bound follow the variations in the actual porosity. In Figure 22(b) the actual brine porosity for the Barrow data is compared with the lower inverse bounds  $\phi_1^{\ell}$  and  $\phi_2^{\ell}$ . The inverse bounds are obeyed, and the variations in  $\phi(t)$  are reflected in the reconstructions.

### Measurements of Fluid and Electrical Transport in Antarctic Sea Ice

During September and October of 2007, Adam Gully (University of Utah math student) and I participated in a six-week voyage into the Antarctic sea ice pack. The Sea Ice Physics and Ecosystem eXperiment (SIPEX), conducted off the coast of East Antarctica, was a major component of Australia's project contributions to the worldwide International Polar Year activities of 2007–2008. An extensive science program conducted by fortyfive scientists from ten nations was led jointly by the Antarctic Climate & Ecosystems Cooperative Research Centre (ACE CRC) and the Australian Antarctic Division. Valuable information was obtained that will help develop more accurate global climate models and will also help us understand how polar ecosystems might respond to future climate change.

Researchers employed a suite of cutting-edge technologies, including airborne laser altimetry and radar, and an under-ice remotely operated vehicle (ROV). For the first time ever in Antarctica, helicopter flights covered thousands of kilometers tracking over the sea ice, gathering large-scale information about the height of the snow and ice surfaces above sea level. The helicopter data were complemented by information gathered from ice coring and other work directly on the ice. The combined information will help validate satellite altimetry data, which will eventually be used to monitor changes in sea ice thickness around Antarctica.

Biologists combined classical ice coring techniques with the ROV to study the underside of the sea ice to map the distribution of sea ice algae. The ROV observations showed high concentrations of Antarctic krill living on the underside of the sea ice as well as in cracks between ice floes. Other scientists worked on ice algae physiology and sea ice biogeochemistry to better understand processes in the Antarctic sea ice zone during the transition from winter to summer. Oceanographers studied the water mass properties and currents beneath the sea ice and found that, contrary to what was expected, the patterns of sea ice drift appear to be affected more by ocean currents than by wind.



normal bone

osteoporotic

Figure 23. Bone porosity can be estimated from torsional modulus data using reconstruction of the spectral measure, similar to estimating brine porosity in sea ice from permittivity data. Such work will help in monitoring osteoporosis [1]. The math doesn't care if it's sea ice or bone!

Gully and I conducted field studies of the fluid and electrical transport properties of first-year sea ice in conjunction with our mathematical modeling work. We made the first fluid permeability measurements of Antarctic pack ice and conducted fluid tracer experiments on extracted blocks of sea ice. In close connection with these fluid measurements, we used surface impedance tomography via a Wenner electrode array to reconstruct the conductivity profile of the ice. We also modified our Wenner configuration to directly measure the vertical conductivity component of extracted cores-a first for sea ice. These measurements will allow the fluid permeability data to be directly related to sea ice electrical properties. Our electrical work, joint with voyage leader Tony Worby, was done in the same exact locations and configurations as EM induction thickness measurements conducted by Japanese sea ice scientist Kazu Tateyama for later correlation. We are still in the process of analyzing the data and preparing it for publication, so our results will be presented elsewhere.

### Conclusions and Future Challenges in Sea Ice Modeling

Numerous challenges remain in the modeling of sea ice for climate applications. Many important processes are only crudely represented, such as snow-ice formation, albedo evolution, aspects of the snow pack overlying the ice cover, and the parameterization of ice strength and dynamic-driven ice ridging and rafting. Figure 25 shows dire model predictions for the decline of the summer Arctic sea ice pack, and much work remains to refine these predictions based on our increasing knowledge of important processes in sea ice and how the pack interacts with the ocean and atmosphere.

Our work on fluid transport is central to a broad range of problems in the geophysics and biology of sea ice, yet fundamental studies have been lacking. We have developed a unified theory of the fluid permeability of sea ice and its dependence on microstructural evolution with temperature. The theory ties together different approaches to estimating effective composite behavior and closely captures laboratory and Arctic field data. We have also made progress in CT imaging of sea-ice microstructure, quantifying the thermal evolution of pore connectivity and morphology at the (sub)granular scale, which in turn has informed the theoretical advances. Our work on sea ice electrical properties helps provide a theoretical basis for new techniques in remote sensing, in particular those which help to monitor the impact of climate change.

Our investigations demonstrate how to link information about the microscale structure to macroscopic transport properties, which in turn constrain global-scale processes of climatological and biological importance. The results yield simple parameterizations of fluid transport through sea ice in terms of porosity  $\phi$ , and therefore the state variables *T* and *S*, and can help to improve the representation of sea ice in climate, biogeochemical, and geophysical models. Future studies will have to examine how these findings impact the simulation of heat and fluid flow through sea ice in large-scale models, which currently oversimplify such processes.

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Figure 24. The author taking a sea ice core in (a) and making a permeability measurement in (b) by measuring the rate at which sea water fills in the partial hole through the ice. (c) Adam Gully (right) sets up a Wenner array, while Kazu Tateyama takes thickness measurements with an EM induction instrument. (d) The Wenner array was adapted to take direct measurements of the vertical conductivity of extracted sea ice cores.

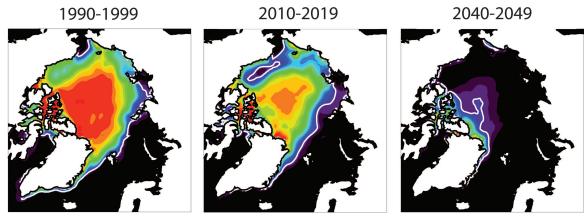


Figure 25. Coupled climate model simulations from the Community Climate System Model, version 3, showing September sea ice concentration averaged for three different time periods: (left to right) 1990-1999, 2010-2019, and 2040-2049. The white line indicates the simulated ice extent, defined as the 15% ice concentration contour. This and other similar climate models suggest significant sea ice loss will occur over the next fifty years, and near ice-free September conditions may be reached in the Arctic by the mid-twenty-first century. Some scientists predict such a decline over an even shorter period of time.



Figure 26. Antarctica is not only of great scientific interest, but is a place of unique beauty and exciting adventures.

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**To find out more:** If you are interested in finding out more about math and sea ice, you can visit the author's webpage, http://www.math.utah.edu/  $\sim$ golden. There you will also be able to link to a video on the SIPEX expedition to Antarctica produced by the author, located under http://www.math.utah.edu/~golden/media. html. A list of additional references for this article is located under http://www.math.utah.edu/ ~golden/publications.html. If you are interested in learning about how math is being used to address the big questions in climate science, including more about sea ice, you are encouraged to visit the Math Awareness Month April 2009 website at http://www.mathaware.org on "Mathematics and Climate".

#### References

- [1] C. BONIFASI-LISTA and E. CHERKAEV, Identification of bone microstructure from effective complex modulus, In *Vibration Problems*, E. Inan and A. Kiris, editors, Springer Proceedings in Physics, volume 111, pages 91–96, 2005.
- [2] F. D. CARSEY, editor, *Microwave Remote Sensing* of Sea Ice, Geophysical Monograph 68, American Geophysical Union, Washington DC, 1992.
- [3] J. FREITAG, The hydraulic properties of Arctic sea ice—Implications for the small-scale particle transport, *Ber. Polarforsch.*, 325:150 pp., 1999.
- [4] K. GOLDEN, Convexity and exponent inequalities for conduction near percolation, *Phys. Rev. Lett.*, 65(24):2923–2926, 1990.
- [5] K. M. GOLDEN, S. F. ACKLEY, and V. I. LYTLE, The percolation phase transition in sea ice, *Science*, 282:2238–2241, 1998.
- [6] K. M. GOLDEN, D. BORUP, M. CHENEY, E. CHERKAEVA, M. S. DAWSON, K. H. DING, A. K. FUNG, D. ISAACSON, S. A. JOHNSON, A. K. JORDAN, J. A. KONG, R. KWOK, S. V. NGHIEM, R. G. ONSTOTT, J. SYLVESTER, D. P. WINEBRENNER, and I. ZABEL, Inverse electromagnetic scattering models for sea ice, *IEEE Trans. Geosci. Rem. Sens.*, 36(5):1675–1704, 1998.
- [7] K. M. GOLDEN, M. CHENEY, K. H. DING, A. K. FUNG, T. C. GRENFELL, D. ISAACSON, J. A. KONG, S. V. NGHIEM, J. SYLVESTER, and D. P. WINEBRENNER, For-

ward electromagnetic scattering models for sea ice, *IEEE Trans. Geosci. Rem. Sens.*, 36(5):1655-1674, 1998.

- [8] K. M. GOLDEN, H. EICKEN, A. L. HEATON, J. MINER, D. PRINGLE, and J. ZHU, Thermal evolution of permeability and microstructure in sea ice, *Geophys. Res. Lett.*, 34:L16501 (6 pages and issue cover), 2007.
- [9] K. M. GOLDEN, A. L. HEATON, H. EICKEN, and V. I. LYTLE, Void bounds for fluid transport in sea ice, *Mechanics of Materials*, 38:801–817, 2006.
- [10] A. GULLY, L. G. E. BACKSTROM, H. EICKEN, and K. M. GOLDEN, Complex bounds and microstructural recovery from measurements of sea ice permittivity, *Physica B*, 394:357–362, 2007.
- [11] U. HORNUNG, editor, *Homogenization and Porous Media*, Springer, New York, 1997.
- [12] M. O. JEFFRIES, editor, Antarctic Sea Ice: Physical Processes, Interactions and Variability, American Geophysical Union, Washington DC, 1998.
- [13] R. P. KUSY and D. T. TURNER, Electrical resistivity of a polymeric insulator containing segregated metallic particles, *Nature*, 229:58–59, 1971.
- [14] M. P. LIZOTTE and K. R. ARRIGO, editors, Antarctic Sea Ice: Biological Processes, Interactions and Variability, American Geophysical Union, Washington DC, 1998.
- [15] D. LUBIN and R. MASSOM, Polar Remote Sensing, Volume I: Atmosphere and Oceans, Springer-Verlag, New York, 2006.
- [16] D. K. PEROVICH, J. A. RICHTER-MENGE, K. F. JONES, and B. LIGHT, Sunlight, water, and ice: Extreme Arctic sea ice melt during the summer of 2007, *Geophys. Res. Lett.*, 35:L11501, 2008.
- [17] M. C. SERREZE, M. M. HOLLAND, and J. STROEVE, Perspectives on the Arctic's shrinking sea-ice cover, *Science*, 315:1533–1536, 2007.
- [18] D. STAUFFER and A. AHARONY, Introduction to Percolation Theory, Second Edition, Taylor and Francis Ltd., London, 1992.
- [19] D. N. THOMAS and G. S. DIECKMANN, editors, Sea Ice: An Introduction to Its Physics, Chemistry, Biology and Geology, Blackwell, Oxford, 2003.
- [20] S. TORQUATO, Random Heterogeneous Materials: Microstructure and Macroscopic Properties, Springer-Verlag, New York, 2002.
- [21] S. TORQUATO and D. C. PHAM, Optimal bounds on the trapping constant and permeability of porous media, *Phys. Rev. Lett.*, 92:255505:1-4, 2004.
- [22] P. WADHAMS, *Ice in the Ocean*, Gordon and Breach Science Publishers, Amsterdam, 2000.
- [23] W. F. WEEKS and S. F. ACKLEY, The growth, structure and properties of sea ice, In *The Geophysics of Sea Ice*, N. Untersteiner, editor, pages 9–164. Plenum Press, New York, 1986.
- [24] P. WONG, J. KOPLICK, and J. P. TOMANIC, Conductivity and permeability of rocks, *Phys. Rev. B*, 30:6606–6614, 1984.
- [25] A. P. WORBY, P. W. GRIFFIN, V. I. LYTLE, and R. A. MASSOM, On the use of electromagnetic induction sounding to determine winter and spring sea ice thickness in the Antarctic, *Cold Reg. Sci. Tech.*, 29:49–58, 1999.
- [26] J. ZHU, A. JABINI, K. M. GOLDEN, H. EICKEN, and M. MORRIS, A network model for fluid transport in sea ice, *Ann. Glaciol.*, 44:129–133, 2006.

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